



Course syllabus

ATLANTA METROPOLITAN STATE COLLEGE

School of Arts and Sciences

Principles of Chemistry and Lab I – CHEM 1211K CRN: 80158 Section 101

TR 11:00 AM – 12:20 PM, ASB (McMillan Building, #900) Room 262

T 2:10 PM – 5:10 PM, ASB (McMillan Building, #900) Room 244

Main Campus Course

This is a Core IMPACTS course that is part of the STEM area.

Core IMPACTS refers to the core curriculum, which provides students with essential knowledge in foundational academic areas. This course will help master course content, and support students' broad academic and career goals.

*****Please download this document as it may not display accurately in the web browser**

	Instructor	Kenneth Starks, MS, MEd
	Office location:	ASB 207
	Office telephone:	404-756-3789 Please text 470-755-7381
	Email Address	kstarks@atlm.edu
	Class Meeting Times:	Lecture: Tuesday and Thursday 11:00 am – 12:20 pm Building #900 Rm. 262 Lab: Tuesday 2:10 pm – 5:10 pm Building #900 Rm. 244
	<ul style="list-style-type: none"> In Person Office Hours: 	Monday 12:30 pm – 5:30 pm Thursday 12:30 pm – 5:30 pm
	<ul style="list-style-type: none"> Online Office Hours 	Monday 1:00 pm – 3:00 pm with the use of Microsoft TEAMS
	<ul style="list-style-type: none"> By Appointment 	Office hour disclaimer: If you are not able to meet with the instructor during office hours, you may schedule an appointment based on instructor's availability.
	Course Information	The following chapters will be covered in this course: <ul style="list-style-type: none"> Chapter 1: Chemical Foundations Chapter 2: Atoms, Molecules, and Ions Chapter 3: Stoichiometry Chapter 4: Types of Chemical Reaction and Solutions Chapter 5: Gases Chapter 7: Atomic Structure and Periodicity Chapter 8: Ionic Bonding
	Pre-requisites	Math 1111 Or 1113 with a C or better
	Credit Hours	4

Catalog Description	For the student majoring in engineering or science, this course is designed to present the fundamentals of chemistry and demonstrate their application through lecture and problem solving. Topics include: measurements, atomic theory, chemical reactions, equations, stoichiometry, solutions and solution stoichiometry, gases and their properties, quantum theory, the periodic table, bonding theories and oxidation-reduction reaction.
Orienting Question	How do I ask scientific questions or use data, mathematics, or technology to understand the universe?
Career Ready Competencies	Inquiry and Analysis Problem-Solving Teamwork Properties of Some Representative Elements Lab
Course Start and End Date	August 18 until December 6
Course Textbooks	<p>Lecture: General Chemistry 10th edition by Steven S. Zumdahl and Susan A. Zumdahl, Cengage, ISBN 1285188497</p> <p>Lab: Experimental Chemistry, 10th Edition, by Steven S. Zumdahl, Susan A. Zumdahl, and Donald Decoste Cengage Learning, Boston, Massachusetts 02210. ISBN-13: 978-1-305-95745-9.</p> <p>This course is a Day 1 Ready Inclusive Access course. This means that these books have already been paid with your tuition. Click on the Start Here Module in Brightspace (D2L), then click on the ebook lab manual link in order to acquire the lab manual. In order to obtain the textbook for lecture each student must visit the AMSC Bookstore website (https://atlm.textbooktech.com/) to request the textbook to be sent to an address. Once on the website, login using your AMSC student credentials, scroll down to Textbook Lookup, click on Lookup by Student ID, then choose the mail option located by the textbook.</p>
Required Resources	Textbook, Scientific calculator, Bound notebook for lab
Recommended Resources	Lab goggles, Safety lab coat
Core Impact Learning Outcomes	Students will use the scientific method and laboratory procedures or mathematical and computational methods to analyze data, solve problems, and explain natural phenomena.
General Education Learning Outcomes	Apply the steps and tenets of the scientific method.
Program Learning objectives	_____
Course Learning Objectives	Science: A Process for Understanding Nature and Its Changes

		<p>1.2 The Scientific Method</p> <p>Scientific Models</p> <p>1.3 Units of Measurement</p> <p>1.4 Uncertainty in Measurement</p> <p>Precision and Accuracy</p> <p>1.5 Significant Figures and Calculations</p> <p>1.6 Learning to Solve Problems Systematically</p> <p>1.7 Dimensional Analysis</p> <p>1.8 Temperature</p> <p>1.9 Density</p> <p>1.10 Classification of Matter</p> <p>TheEarlyHistoryofChemistry</p> <p>2.2 FundamentalChemicalLaws</p> <p>2.3 Dalton'sAtomicTheory</p> <p>2.4 EarlyExperimentstoCharacterize theAtom</p> <p>TheElectron</p> <p>Radioactivity</p> <p>TheNuclearAtom</p> <p>2.5 TheModernViewofAtomic Structure:AnIntroduction</p> <p>2.6 MoleculesandIons</p> <p>2.7 An Introduction to the Periodic Table</p> <p>2.8 NamingSimpleCompounds</p> <p>Binary Ionic Compounds (Type I)</p> <p>FormulasfromNames</p> <p>BinaryIonicCompounds(Type II)</p> <p>IonicCompoundswithPolyatomicIons</p> <p>BinaryCovalentCompounds(Type III)</p> <p>Acids</p>
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	<p>Stoichiometry</p> <p>3.1 Counting by Weighing</p> <p>3.2 Atomic Masses</p> <p>3.3 The Mole</p> <p>3.4 Molar Mass</p> <p>3.5 Learning to Solve Problems</p> <p>3.6 Percent Composition of Compounds</p> <p>3.7 Determining the Formula of a Compound</p> <p>3.8 Chemical Equations</p> <p>Chemical Reactions</p> <p>The Meaning of a Chemical Equation</p> <p>3.9 Balancing Chemical Equations</p> <p>3.10 Stoichiometric Calculations: Amounts of Reactants and Products</p> <p>3.11 The Concept of Limiting Reactant</p> <p>Types of Chemical Reactions and Solution Stoichiometry</p> <p>4.1 Water, the Common Solvent</p> <p>4.2 The Nature of Aqueous Solutions: Strong and Weak Electrolytes</p> <p>Strong Electrolytes</p> <p>Weak Electrolytes</p> <p>Nonelectrolytes</p> <p>4.3 The Composition of Solutions</p> <p>Dilution</p> <p>4.4 Types of Chemical Reactions</p> <p>4.5 Precipitation Reactions</p> <p>4.6 Describing Reactions in Solution</p> <p>4.7 Stoichiometry of Precipitation Reactions</p> <p>4.8 Acid–Base Reactions</p> <p>Acid–Base Titrations</p>
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	<p>4.9 Oxidation–Reduction Reactions Oxidation States The Characteristics of Oxidation– Reduction Reactions 4.10 Balancing Oxidation–Reduction Equations</p> <p>Gases 5.1 Pressure Units of Pressure 5.2 The Gas Laws of Boyle, Charles, and Avogadro Boyle’s Law Charles’s Law Avogadro’s Law 5.3 The Ideal Gas Law 5.4 Gas Stoichiometry Molar Mass of a Gas 5.5 Dalton’s Law of Partial Pressures Collecting a Gas over Water 5.6 The Kinetic Molecular Theory of Gases Pressure and Volume (Boyle’s Law) Pressure and Temperature Volume and Temperature (Charles’s Law) Volume and Number of Moles (Avogadro’s Law) Mixture of Gases (Dalton’s Law) Deriving the Ideal Gas Law The Meaning of Temperature</p> <p>Atomic Structure and Periodicity 7.1 Electromagnetic Radiation 7.2 The Nature of Matter The Photoelectric Effect 7.3 The Atomic Spectrum of Hydrogen</p>
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	<p>7.4 The Bohr Model</p> <p>7.5 The Quantum Mechanical Model of the Atom</p> <p>The Physical Meaning of a Wave Function</p> <p>7.6 Quantum Numbers</p> <p>7.7 Orbital Shapes and Energies</p> <p>7.8 Electron Spin and the Pauli Principle</p> <p>7.9 Polyelectronic Atoms</p> <p>7.10 The History of the Periodic Table</p> <p>7.11 The Aufbau Principle and the Periodic Table</p> <p>7.12 Periodic Trends in Atomic Properties</p> <p>Ionization Energy</p> <p>Electron Affinity</p> <p>Atomic Radius</p> <p>7.13 The Properties of a Group: The Alkali Metals</p> <p>Information Contained in the Periodic Table</p> <p>The Alkali Metals</p> <p>Bonding: General Concepts</p> <p>8.1 Types of Chemical Bonds</p> <p>8.2 Electronegativity</p> <p>8.3 Bond Polarity and Dipole Moments</p> <p>8.4 Ions: Electron Configurations and Sizes</p> <p>Predicting Formulas of Ionic Compounds</p> <p>Sizes of Ions</p> <p>8.5 Energy Effects in Binary Ionic Compounds</p> <p>Lattice Energy Calculations</p>
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	Enrollment Status:	Students are ultimately responsible for ensuring that the course(s) in which they enroll are included in the approved degree plan and program map for their program of study. Students must periodically check their enrollment status in this course during the semester. The student is responsible for determining changes, if any, in enrollment status and taking necessary steps (e.g., pursuing re-instatement in this course) following those outlined in the AMSC catalog.
	Attendance Verification (No Show)/Reinstatement	<p>Atlanta Metropolitan State College has a "No-Show" Reporting (Attendance Verification) policy. This policy is to comply with Federal Financial Aid regulations. Financial Aid recipients at Atlanta Metropolitan State College may become ineligible for funds by not attending class session (per enrolled course). Students who do not complete Mandatory Attendance Assignments and attend class sessions are NOT entitled to keep their financial aid award. The Registrar's Office will notify the students and faculty when the Attendance Verification Period has opened. The established "No-Show" Reporting (Attendance Verification) procedure will enable Atlanta Metropolitan State College to adjust financial aid awards before funds are issued to students (thereby eliminating liability for both the College and the student). A student reported as non-attending a course must seek the approval of the instructor in order to be reinstated. Once approved, the student will complete the Reinstatement form and submit it. The Office of Registrar will notify students when course reinstatement process has been completed during Reinstatement Period.</p> <p>The student can demonstrate compelling reason (s) that have prevented attendance and the instructor believes that there is a strong probability that the student can catch up in the class OR The instructor made an error, and the student was in attendance prior to being dropped.</p>
	Computer Hardware & Software Requirements:	One of the challenges many encounter with enrolling in and completing an online course is the accessibility to the required and recommended software and hardware. Different institutions, and even different courses within the same institution, have varying technology requirements. Check your hardware and software systems to determine its compatibility with the online course you have selected.
	Minimum Hardware Recommendations to take courses:	<p>D2L System Requirement Internet Connection:</p> <ul style="list-style-type: none"> · Ethernet Network Capability required · Wireless Network Capability required <p>Operating System Requirements</p> <ul style="list-style-type: none"> • PC: <ul style="list-style-type: none"> • Compatible Operating System: • Web Browser: Firefox, Chrome

		<ul style="list-style-type: none"> • Mac: <ul style="list-style-type: none"> • Compatible Operating System: • Web Browser: Firefox, Chrome, Safari <p>Hardware requirements:</p> <p>Minimum Technical Specifications for Hardware:</p> <ul style="list-style-type: none"> • A processor of 2GHz or faster • 4GB RAM or greater • 500 GB of Hard Drive space • Monitor and video card with a minimum resolution of 1024x768 • Keyboard and mouse <p>Minimum Technical Specifications for Computer Peripherals:</p> <ul style="list-style-type: none"> • Speakers • Headphones • Microphone • Webcam <p>Software requirements:</p> <p>Browser Requirements/Supported Browsers</p> <p>Compatible Browsers:</p> <ul style="list-style-type: none"> • Apple Safari - https://support.apple.com/downloads/safari (Mac) • Google Chrome - https://www.google.com/chrome/ (Mac or PC) • Mozilla Firefox - https://www.mozilla.org/en-US/firefox/new/ (Mac or PC) <p>Application Software</p> <ul style="list-style-type: none"> • Microsoft Office 2016 (Word, Excel, PowerPoint) (Mac or PC) • Adobe Reader - https://get.adobe.com/reader/ <p>Plug-ins</p> <ul style="list-style-type: none"> • Java - https://www.java.com/en/download/ • Adobe Flash Player - https://get.adobe.com/flashplayer/ • Windows Media Player - https://www.microsoft.com/en-us/download/details.aspx?id=20426 • Apple QuickTime - https://support.apple.com/downloads/%2523quicktime • Microsoft Silverlight - https://www.microsoft.com/getsilverlight/Get-Started/Install/Default <p><u>Accessibility:</u> Wi-Fi is also available for use in the campus parking lots.</p>
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Tutoring Services	<p>Students improve their self-confidence and increase their chances of excelling in their courses when they utilize their college/university academic support services. The following tutoring services are available to AMSC students:</p> <ul style="list-style-type: none"> ❖ The AMSC Writing Center is open year-round to support students, staff, and faculty at AMSC. It offers virtual and in-person tutoring for various writing projects, assists at any stage of the writing process, and increases improvement in structure, use of sources, style, grammar, and more. The Writing Center is located in Building 100, Room 211. Students can walk in during hours of operation or schedule an appointment at https://calendly.com/amscwritingcenter/30min ❖ Tutor.com is a virtual space for students to access writing tutors outside of AMSC faculty. Access tutor.com in Brightspace by logging into your course. Next, click the 'Free Tutoring' tab and choose Tutor.com from the dropdown options. Students are granted five (5) hours per semester, and registration is not required. Contact the Center for Student Success and Advising at (404) 756-5690 for assistance. ❖ TutorOcean is a tutorial for students enrolled in STEM courses such as Biology, Chemistry, Mathematics, and Computer Science. Access TutorOcean and sign up at https://atlm.tutorocean.com.
Americans with Disabilities Act (ADA) Statement	<p>Atlanta Metropolitan College is committed to providing support for all students and making their college experience an enriching opportunity. In compliance with Section 504 of the Rehabilitation Act of 1973, and the Americans with Disabilities Act of 1990, The Department/Office of Counseling and Accessibility Services, located in the Student Services & Success Center, building 650-Suite 252, oversees the coordination of services for students with documented disabilities. The Coordinator of Disability Services collaborates with faculty and staff to offer provisions for reasonable accommodation to students who meet the requirements.</p> <p>It is the policy and practice of AMSC to make all Web information accessible to students with disabilities. If you, as a student with a disability, have difficulty accessing any part of the course materials for this class, please notify the instructor immediately.</p> <p>Accommodation cannot be provided until a reasonable accommodation plan is in place. To the greatest extent possible, all college representatives shall observe confidentiality.</p>

Office of Counseling and Accessibility Services	<p>The Office of Counseling and Accessibility Services operates under the Americans with Disabilities Act (ADA) laws in order to assist in leveling the playing field for students who have disabilities with those who do not.</p> <p>The amended ADA, otherwise known as ADAAAA defines “disability” as a physical or mental impairment that substantially limits one or more major life activities. If you feel that you have a disability or impairment that may limit your academic functioning, please contact Dr. Dorothy Williams, the Director of Counseling and Accessibility Services at 404-756-4016 or at</p> <p>https://www.atlm.edu/students/counseling-and-disability-services.aspx.</p> <p>The Coordinator of Counseling and Accessibility Services reviews all accommodation requests. In order to receive accommodation, the student’s illness or disability must be verified in writing by a physician, psychiatrist, or some other health care provider or specialist. Students choosing to access disability support services should contact the Coordinator as soon as possible after acceptance to AMSC. Please be aware that late notifications may result in complications for establishing accommodation in a timely fashion.</p>
Withdrawal	<p>Withdrawal from a course is solely the responsibility of the student. Instructors will not initiate student withdrawals. A student who wishes to withdraw from a course MUST submit a completed Withdrawal Form (Schedule Reductions Form) to the Registrar’s Office before mid-term in order to receive a grade of “W” for the course.</p> <p>A student who withdraws after the Midterm date receives a “WF” unless the Vice President for Academic Affairs determines that it is a hardship case, then a “W” will be recorded. The possibility that a student may fail the course will not be considered a hardship.</p>
Incomplete Grade Policy	<p>An incomplete may be awarded at the instructor's discretion for non-academic reasons which prevent the student from completing the course requirements. The student must be passing the course at the time that the Incomplete is awarded and must sign an "Awarding of Incomplete" agreement. Unless otherwise stated, the incomplete should be removed by the end of the following semester; otherwise, the instructor will change the grade to an “F” grade.</p>
Time Commitment	<p>To successfully complete this course it requires discipline, devoted time and commitment. A student must arrange his / her schedule to allow for the required time for this course. Expect to spend a minimum of 9 hours per week.</p>

Student Expectations	You are expected to refrain from profanity, crudeness, and slurs of any kind. You are also expected to treat your fellow students and instructor just as you would in the traditional classroom. Proper conduct applies to all forms of communication in the course. The penalty for not complying with these guidelines is removal from the course. Only the epitome of professionalism is expected of each student.
Extra Credit	Provide proof of attendance of tutor ocean or tutor.com study sessions. Three extra credit points will be awarded on the lowest quiz for each proof of a half hour workshop attended. GELO Quiz, points will be added to the lowest quiz based on the number of questions answered correctly, maximum 10 points.
Online Discussion Protocol	This class is taught face-to-face.
Degree Relevance and Enrollment Status	Students are ultimately responsible for ensuring that the course(s) in which they enroll are included in the approved degree plan and program map for their program of study. Students must periodically check their enrollment status in this course during the semester. The student is responsible for determining changes, if any in enrollment status and taking necessary steps (e.g. pursuing re-instatement in this course) following those outlined in the AMSC catalog
Abandoning a Course	Abandoning a course should be avoided at all cost. Abandoning a course instead of following official drop procedures will result in a grade of —F at the end of the course. It is the student's responsibility to initiate and complete the withdrawal process.
Academic Honesty	<p>Only the epitome of professionalism is expected of each student. Cheating or the abetment of cheating is not tolerated.</p> <p>Per page 91-93 of the 2019-2020 Atlanta Metropolitan State College Catalogue the Penalties for Academic Misconduct states:</p> <ul style="list-style-type: none"> • In cases where a student is found guilty of cheating or exhibiting academic misconduct involving an instructor-generated assignment or examination, the instructor may impose a penalty. • Types of penalties may include, but are not limited to, the instructor assigning a grade of "F" for the assignment, the instructor not accepting the work, the student being assigned additional work, or the student receiving a grade reduction for the assignment. <p>The maximum penalty the instructor may impose is a grade of "F" for the course. The instructor will impose the maximum penalty in this class in the case of academic dishonesty.</p>
Class Cancellation	Procedure regarding long-term emergency closure of the college (attendance policy): In the event of an emergency that forces the college to close for an extended period, students MUST contact the instructor of this class within 48 hours using the contact information (e.g., email address in

		<p>BrightSpace/D2L) on the syllabus to obtain directions for continuing the course. The instructor will provide directions for the transmission and submission of course assignments and course assessments, including due dates.</p> <p>The student is responsible for submitting valid, accurate contact information, including an active AMSC email address to the instructor by the end of the first week of the course. Students can obtain an Atlanta Metropolitan State College Student email address in the Academic Support Center on the third floor of the Library Building.</p> <p>If the instructor for the course cannot be reached within the specified period (within 48 hours), the Dean of the School responsible for the course can be reached at the email address posted on the college's website</p>		
	Grade Scale	<p>Grading Scale: 100% - 90% = A 89% - 80% = B 79% - 70% = C 69% - 60% = D Below 60%= F</p>		
	Class Schedule	Scales and Weights		
	These scheduled dates and assignments may be subject to change at the discretion of the instructor.	Category	Weight	Assignment/Due Date
		Exams	30%	Exam 1 Chapters 1-2 9/23
				Exam 2 Chapters 3-5 10/30
				Exam 3 Chapters 7-8 12/4
		Quizzes	20%	Syllabus 8/25
				Quiz 1 9/4

				Quiz 2 10/9
				Quiz 3 11/25
		Lab Exercises	25%	See Lab
		Homework	10%	Weekly
		Scientific Method	Tenets of the scientific method tutorial	NA
		GELO Extra Credit Assessment		TBA
		Final	15%	12-5/11
	Grade Appeals and Student Complaint Policy and Process	<p>Please follow the Grade Appeals Process outlined in the AMSC Student Catalog, Pages 61/62.</p> <p>You can also refer to the Grade Appeal brochure at: https://www.atlm.edu/downloads/advisement/CAAS%20Grade%20Appeal%20Brochure.pdf</p> <p>For student complaint policy and process, refer to AMSC student catalog pages 51/52.</p>		
	Frequently Asked Questions and Helpful Links			
	I need:	LINK		
	What is Brightspace (D2L) and how can I access it?	<p>Brightspace(D2L) is the virtual space where students access their online courses and some Face2Face class resources, quizzes, assignments, etc.</p> <p>You can access Brightspace (D2L) from https://atlm.view.usg.edu/</p> <p>You can also access Brightspace (D2L) from the College's webpage and click on Brightspace (D2L) on top.</p>		
	Who is my Advisor? Where can I receive Advisement and Tutoring assistance?	Center for Academic Advising and Success (CAAS)		
	What do I do if I face technical issues while taking a	https://d2lhelp.view.usg.edu/ You can reach the GaView Helpdesk 24/7/365 days at 18557724423		

quiz or turning in an assignment in Brightspace	If you are unable to submit a quiz or assignment or face a technical glitch, please contact the University System of Georgia's GaView Helpdesk at 1855 772 4423. The Helpdesk is open 24/7 all 365 days. If your issue cannot be resolved right away, the Helpdesk will issue a ticket to your Atlm Ginger email address. That ticket needs to be forwarded to your instructor to prove that you faced a technical issue that forced you to miss a deadline.
I can't download Respondus Lockdown Browser. What do I do?	If you are unable to download Respondus to your computer, please email the Office of Testing at Testing@atlm.edu at least 24 hours in advance requesting a testing appointment at the Testing Lab. For Respondus issues, please contact https://web.respondus.com/contact/ .
Where do I go for ADA Accommodations	https://www.atlm.edu/students/counseling-and-disability-services.aspx
I have a complaint. Where do I go?	Fill out and submit the Student Complaint form
I have an issue with my grade. How may I appeal my grade.	Read the brochure and follow the instructions to appeal your grade .
How do I Withdraw from a Course	Follow the Course Withdrawal Process here
I have a Hardship. How do I do a Hardship Withdrawal?	Follow the Hardship Withdrawal Process here
To Know the Campus Carry/HB 280 Policy	House Bill 280 Guidelines
Disclaimer	<i>Information contained in this syllabus and schedule was, to the best knowledge of the instructor, considered correct and complete when distributed for use at the beginning of the semester. This syllabus should be considered only a guide for instructor and students, not a formal contract between Atlanta Metropolitan State College and any student. The instructor reserves the right, acting within the policies and procedures of AMSC, to make changes in course content or instructional techniques.</i>

